

THURSDAY, DECEMBER 14, 2017 AP CHEMISTRY
CHAPTER 21 SUPPLEMENTAL NOTES

WHAT IS A LOGARITHM?

IT IS A WAY OF ASKING, WHAT POWER OF A CERTAIN NUMBER (CALLED THE BASE OF THE LOGARITHM) GIVES THE NUMBER "INSIDE" THE LOGARITHM?

$$\begin{array}{lll} \log_{10} 1 = 0 & \log_2 1 = 0 & \text{power}_7 7 = 1 \\ \log_{10} 10 = 1 & \log_2 2 = 1 & \text{power}_7 49 = 2 \\ \log_{10} 100 = 2 & \log_2 4 = 2 & \text{power}_7 343 = 3 \text{ ETC.} \end{array}$$

WHAT ARE THEY FOR?

LOGARITHMS ARE DESIGNED TO TAKE A HARD PROBLEM (MULTIPLYING LARGE NUMBERS) AND MAKE IT AN EASY ONE (ADDING SMALL NUMBERS).

$$\textcircled{1} 10^7 \times 10^5 = ? \quad \textcircled{2} \log_{10}(10^7 \times 10^5) = \log_{10} 10^5 + \log_{10} 10^7$$

$$\textcircled{3} 5 + 7 = 12 \quad \textcircled{4} 10^{12} \text{ (OR, ANTILOG } 12)$$

$$\begin{array}{l} 57,000,000,000 \times 42,000,000 \\ \log_{10} 57,000,000,000 + \log_{10} 42,000,000 \\ 10.7558 + 7.6232 = 18.379 \\ 10^{18.379} = 2,390,000,000,000,000,000 \end{array}$$

THINK OF LOGARITHMS AS THE REVERSE OF EXPONENTIATION

$$a + b = c \quad \text{REVERSE: } a = c - b$$

$$a \cdot b = c \quad \text{REVERSE: } a = c \div b$$

$$a^b = c \quad \text{REVERSE: } \log_a a^b = b = \log_a c$$

①

PROPERTIES OF LOGARITHMS

$$\log(M \times N) = \log M + \log N \quad (a^b \cdot a^c = a^{b+c})$$

$$\log(M \div N) = \log M - \log N \quad \left(\frac{a^b}{a^c} = a^{b-c}\right)$$

$$\log(M^x) = x \log M \quad ((a^b)^c = a^{b \cdot c})$$

$$a^{\log_a M} = M$$

ON YOUR CALCULATOR YOU HAVE TWO LOGARITHM BUTTONS.

① log OR LOG MEANS \log_{10}

(ANTILOG LOOKS LIKE 10^x)

② ln MEANS \log_e WHERE e IS A
MATHEMATICAL CONSTANT
 $e \cong 2.71828 \dots$

FOR EXPONENTIAL DECAY, THE ALGEBRAIC EXPRESSION IS
GIVEN IN TERMS OF e : $y = e^{-x}$

THIS IS WHY WE USE \ln , ALSO CALLED THE
NATURAL LOGARITHM: $\ln y = -x$

FOR RADIOACTIVE DECAY PROBLEMS WE USE \ln .

$$\frac{N}{N_0} = e^{-kt} \quad \ln\left(\frac{N}{N_0}\right) = -kt$$

CHAPTER 21 SUPPLEMENTAL NOTES, CONTINUED.

RATES OF DECAY - DERIVING THE EQUATION

IN NUCLEAR DECAY A MEASURED RATE (AS IN, DECAYS PER SECOND) IS PROPORTIONAL TO THE NUMBER OF ATOMS (N).

$$\text{RATE} = -kN$$

$$\text{OR}$$

$$\frac{\Delta N}{\Delta t} = -kN$$

OR, IN THE LANGUAGE OF CALCULUS

$$\frac{dN}{dt} = -kN$$

SINCE THE ATOMS ARE REDUCING THEIR NUMBERS OVER TIME, THE EXPRESSION INCLUDES A NEGATIVE SIGN.

THE EQUATION $\frac{dN}{dt} = -kN$ TELLS US HOW RATE CHANGES AS A FUNCTION OF THE NUMBER OF ATOMS.

WE WANT TO KNOW HOW THE NUMBER OF ATOMS CHANGES AS A FUNCTION OF TIME. INTEGRAL CALCULUS PROVIDES THAT FUNCTION.

$$\textcircled{1} \frac{dN}{dt} = -kN$$

$$\textcircled{2} \frac{1}{N} dN = -k dt$$

$\textcircled{3}$ DEFINITE INTEGRAL

$$\int_{N_0}^N \frac{1}{N} dN = - \int_0^t k dt$$

$$\textcircled{4} \ln(N) - \ln(N_0) = -kt$$

$$\textcircled{5} \ln\left(\frac{N}{N_0}\right) = -kt$$

$$\textcircled{6} \frac{N}{N_0} = e^{-kt}$$

N/N_0 IS THE DECIMAL FRACTION OF ATOMS THAT REMAIN AT TIME t .

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